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joven_diosa. Terms in this set (10)
Kinetic theory. 1. All matter is composed of small particles (Atoms, molecules and ions). 2. These particles are in constant, random motion. 3. These particles are colliding with each other and the walls of their containers.

~~Chapter 16 section 1 solids, liquids & gases(kinetic ...~~

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17. The kinetic energy Of a substance is kinetic energy of its particles. 18. to change from a liquid to a gas at temperatures above those normal to the liquid state 19. process that occurs during boiling 1. to change from solid to liquid 2. energy needed to change a material from liquid to gas (3 words)

Quia

What are the 3 assumption of the Kinetic Theory? (1) all matter is composed of small particles (atoms, molecules and ions), (2) particles are in constant motion and (3) the particles are colliding with each other and walls of their container What is thermal energy? the total energy of the material's particles (kinetic vibrations, etc)

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KE = $\frac{1}{2}m \times V^2$) = 4 m/s² . What is the amount of work done when a force of 10N moves a 20kg mass of 8 meters? 80J (Work = force X distance) What is the kinetic energy of a bicycle with a mass of 25kg traveling at a velocity of 4m/s? (KE = $\frac{1}{2}m \times V^2$) = 200J. A force of 15N is used to push a box along the floor of 3 meters. How much work is done?

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~~answer key~~

Near the Earth's surface air becomes heated, expands, and rises then as it rises it cools, contracts, and sinks. This creates convection currents in the atmosphere.

~~Section 11.1~~ □ ~~Atmosphere Basics~~

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The kinetic molecular theory describes the behavior of gases in terms of particles in. 2. The kinetic-molecular theory makes the following assumptions.a.In a sample of a gas, the volume of the gas particles themselves is very compared to the volume of the sample.

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Chapter 13 Study Guide Chapter

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~~Summary - Section 1 - What is energy?~~

1. Energy is the ability to cause change. 2. A moving object has kinetic energy that depends on the object's mass and speed. 3. Potential Energy is energy due to position and depends on an object's mass and height. 4.

~~Chapter 13 Study Guide Energy and Energy Resources~~

About This Quiz & Worksheet. The questions on the quiz will test you on the various definitions of terms related to the kinetic theory of matter.

~~Quiz & Worksheet - The Kinetic Theory of Matter | Study.com~~

1. Gases consist of large numbers of tiny particles that are far apart relative to their size. 2. Collisions between gas particles and between particles and container walls are elastic collisions. 3.

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Gas particles are in common, rapid, random motion. They therefore possess kinetic energy, which is energy of motion. 4.

~~Chapter 10.1: States of Matter – Gases~~ ~~– Study Guide ...~~

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