

Phet Photoelectric Effect Answers

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Visualize and describe the photoelectric effect experiment. Correctly predict the results of experiments of the photoelectric effect: e.g. how changing the intensity of light will affect the current and the energy of electrons, how changing the wavelength of light will affect the current and the energy of electrons, how changing the voltage of ...

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List the answers below. From your knowledge of light energy, electrons, wavelength, and frequency, predict the graph of the electron energy as function of wavelength by sketching the graph below. ... Photoelectric Effect with PhET simulations ...

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~~Part II – The Photoelectric Effect. (1) When the light shines on the surface of a metal, the light may eject electrons from the metal with additional kinetic energy. To determine how much kinetic energy is being given to the electrons from the photons of light, a set of parallel plates was setup in a vacuum.~~

~~Lab Photoelectric Effect PhET simulation key~~

~~Photoelectric Effect PhET Lab . Name: _____ Open the Photoelectric Effect PhET Simulation. **Before adjusting anything answer the questions below** 1) Before you begin, what do will happen to ... a) the metal surface when light strikes it . b) the light the intensity slider is moved~~

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Phet Lab Photoelectric Effect Answers Correctly predict the results of experiments of the photoelectric effect: e.g. how changing the intensity of light will affect the current and the energy of electrons, how changing the wavelength of light will affect the current and the energy of electrons, how changing the voltage of light will affect the current and the

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Important modeling notes / simplifications: used in the PHET Photoelectric Effect Computer Simulation* • Electrons are emitted with a range of energies because photons can eject electrons with a range of binding energies. If more of a photon's energy is used to release an electron, the emitted electron will have less kinetic energy.

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Answers Phet Photoelectric effect. 1. Determine the minimum frequency required to have Photoelectric effect for an EM radiation, when incident on a zinc metal surface. 2. Determine the target material if the threshold frequency of EM radiation is 5.5×10^{15} Hz in a particular photoelectric experimental set up. 3. Photoelectric

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Student Exploration Photoelectric Effect Answers The photoelectric effect occurs because the electrons at the surface of the metal tend to absorb energy from the incident light and use it to overcome the attractive forces that bind them to the metallic nuclei. An illustration detailing the emission of photoelectrons as a result of the

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This Student Guide, designed by HS teacher Jonathan Carlson for the calculus-based or AP physics course, takes students step-by-step through the calculations associated with the photoelectric effect. Students will use the PhET Photoelectric Effect simulation to calculate stopping voltage, kinetic energy of a photon in Joules, and learn how to determine threshold frequency and work function from graphical data. This resource contains an answer key.

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