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ORGANIC CHEMISTRY: SOME BASIC  
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(CH\_20)How To Get an A in Organic  
Chemistry How to Memorize Organic  
Chemistry Reactions and Reagents

[Workshop Recording] 01 - Introduction

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To Chemistry - Online Chemistry Course -  
Learn Chemistry \u0026amp; Solve Problems  
SN1, SN2, E1, \u0026amp; E2 Reaction  
Mechanism Made Easy! Organic II - Final  
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chemistry mechanisms - revision Periodic  
Table Explained: Introduction Sigma and Pi  
Bonds: Hybridization Explained!

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~~Nomenclature: Functional groups~~

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1.1: The Origins of Organic Chemistry The term "organic chemistry" was first used in the 1800's to distinguish reactions performed by chemists from those occurring within living organisms.

1.2: Principles of Atomic Structure (Review) Atomic structure is applied to several topics of organic chemistry.

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1: Introduction and Review - Chemistry  
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Introduction and Review 1 17) Provide a

Lewis structure for a molecule with

molecular formula  $\text{CH}_2\text{O}_2$ . Answer: Diff:

2 Section: 1.4 18) Draw the Lewis structure

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for 2-butanol,  $\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CH}_3$ .

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1 Introduction and Review 1) While you  
were up late one night studying organic

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chemistry, you happened to see the last 5 minutes of an infomercial on TV. The spokesperson claimed that their brand of automobile tires were superior to

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professors of science faculties universities.

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ORGANIC CHEMISTRY I CHEM 2323

Fall 2014 Instructor: Michael C. Biewer

biewerm@utdallas.edu, BE 3.326 Syllabus.

Lecture Notes: The following material can  
be downloaded as pdf files. Material for test  
#1 . Chapter 1 Chapter 2 Chapter 3 Chapter

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5 . Material for test #2 . Chapter 4

Organic Chemistry I

Chapter 1 Wade Organic Chemistry. acid-dissociation constant ( $K_a$ )  $pK_a$ . Arrhenius acid. Arrhenius base. the equilibrium constant for the reaction of the acid with wat.... the negative logarithm of  $K_a$ .



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dissociates in water to give  $\text{H}^+$  and  $\text{O}^{2-}$ .

dissociates in water to give  $\text{OH}^-$ .

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Simek - StuDocu

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Structure and Properties of Organic

Molecules 1) An orbital can be described by  
its \_\_\_\_\_, which is the mathematical

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description of the shape of the electron wave as it oscillates. Answer: wave function Diff: 1 Section: 2.1 2) The electron density at any point is proportional to the \_\_\_\_\_ of the electron ...

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7th Edition L. G. Wade, Jr. Infrared Spectroscopy and Mass Spectrometry 2.  
Chapter 12 2 Introduction • Spectroscopy is a technique used to determine the structure of a compound.

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Origins of Organic Chemistry 1.2 Principles  
of Atomic Structure 1.3 Bond Formation:  
The Octet Rule 1.4 Lewis Structures 1.5  
Multiple Bonding 1.6 Electronegativity and

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Bond Polarity 1.7 Formal Charges 1.8 Ionic Structures 1.9 Resonance 1.10 Structural Formulas 1.11 Molecular Formulas and Empirical Formulas

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1. 1 Organic Chemistry, 7e (Wade) Chapter  
14 Ethers, Epoxides, and Sulfides 1) What is  
the hybridization of the oxygen atom in  
dialkyl ethers? A)  $sp^3$  B)  $sp^2$  C)  $sp$  D)  $s$  E)  $p$

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Answer: A Diff: 1 Section: 14.2 2) Which of the following corresponds to the COC bond angle in dimethyl ether? A)  $60^\circ$  B)  $94^\circ$  C)  $110^\circ$  D)  $122^\circ$  E)  $180^\circ$  Answer: C Diff: 1

...

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Technology. Course. Organic Chemistry I  
(CHEM 243) Book title Organic Chemistry;  
Author. LeRoy G. Wade; Jan Simek

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Acids and Bases; Functional Groups 1) An

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orbital can be described by its \_\_\_\_\_, which is the mathematical description of the shape of the electron wave as it oscillates. Answer: wave function Diff: 1 Section: 1.12 LO: 2.1

2) The electron density at any point is proportional to the \_\_\_\_\_ of the electron ...



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