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General Information . Molar gas volume =
24.0 dm³ mol⁻¹ at room temperature and
pressure, RTP Avogadro constant, N_A =
6.02 × 10²³ mol⁻¹ Specific heat capacity
of water, c_w = 4.18 J g⁻¹ K⁻¹ Ionic
product of water, K_w = 1.00 × 10⁻¹⁴
mol² dm⁻⁶ at 298 K 1 tonne = 10⁶ g
Arrhenius equation: $k = Ae^{-E_a/RT}$ or $\ln k$

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