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Cathode rays were discovered by William Crooke's & J.J. Thomson (1880).

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subatomic structures

called Protons,

Neutrons, and Electrons.

Electrons are the

smallest of the three

particles that make up

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fundamental chapter of  
studying Chemistry. It is  
thus vital for you to

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develop a stronghold on this chapter so that you ace the subject as a whole. The MCQ on atomic structure is meant to assess your understanding and grasp over the minute details included in the chapter.

**MCQs on Structure of Atom for NEET - Vedantu**

The central nucleus of

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the atom contains the neutrons and positively charged protons. The nucleus is surrounded by as many orbital electrons (each with a charge of  $-1$ ) as there are. protons (each with a charge of  $+1$ ) in the nucleus. The three sub-atomic particles differ from each other in two important ways.

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Chapter 1 page 1-16 -

University of  
Manchester

(a) atomic mass. (b)  
number of neutrons. (c)  
mass number. (d)  
atomic number Answer:

(d) Atoms of the same  
element can differ in (a)  
chemical properties. (b)  
mass number. (c) atomic  
number. (d) number of  
protons and electrons.

Answer: (b) The



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average atomic mass of  
an element is the  
average of the atomic  
masses of its (a)  
naturally occurring ...

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The element  $Z = 114$  has  
been discovered  
recently. It will belong  
to which of the

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following family/group  
and electronic  
configuration? (a)

Halogen family,  $[Rn]5f$   
 $14 6d 10 7s 2 7p 5$  (b)

Carbon family,  $[Rn]5f$   
 $14 6d 10 7s 2 7p 2$  (c)

Oxygen family,  $[Rn]5f$   
 $14 6d 10 7s 2 7p 4$  (d)

Nitrogen family,  $[Rn]5f$   
 $14 6d 10 7s 2 7p 6$

**NEET and AIPMT**

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Structure of Atom

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MCQs on Structure of  
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Everything around us is made up of atoms. Atom is the smallest unit of matter and has all the chemical properties of a matter. Atom has a positively charged nucleus, which contains positively charged particles known as

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particles called  
neutrons. The nucleus is  
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